



Pergamon

Synthesis of Substituted 4(*Z*)-(Methoxyimino)pentyl-1-piperidines as Dual NK₁/NK₂ Inhibitors

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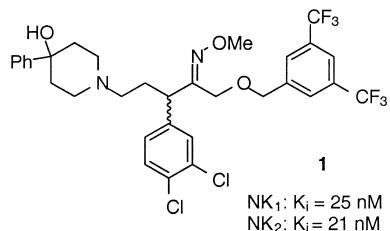
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Abstract—The NK₁ and NK₂ receptor activity of a series of 5-[(3,5-bis(trifluoromethyl)phenyl)methoxy]-3-(3,4-dichlorophenyl)-4(*Z*)-(methoxyimino)pentyl-1-piperidines was evaluated. Compounds **11d**, **11e**, **11f**, **12a**, and **12k** were found to be our most potent inhibitors. © 2001 Elsevier Science Ltd. All rights reserved.

Introduction

The tachykinins are a family of neuropeptides that share a common C-terminal sequence of Phe-X-Gly-Leu-Met-NH₂ and are found throughout the central and peripheral nervous system.¹ There exists three G-protein linked neurokinin receptors (NK₁, NK₂, and NK₃) through which the biological effects are transmitted. Although each neurokinin can act as an agonist at all three receptors, Substance P (SP), Neurokinin A (NKA), and Neurokinin B (NKB) have the highest affinity for the NK₁, NK₂, and NK₃ receptor, respectively.² The neurokinins may play an important role in several disease states, which include migraine, emesis, pain, arthritis, depression, anxiety, and asthma.³ Our research interest lies in the theory that both SP and NKA are responsible for the excessive mucus secretion, airway constriction, and plasma extravasation found in the pathology of asthma.⁴

Our goal was to design and synthesize a compound that will block both the NK₁ and NK₂ receptors as a means of alleviating asthma. Compound **1** was identified as a potent dual NK₁ and NK₂ receptor antagonist.⁵ In this communication, we describe a series of analogues that have diverse substituents at the 4-position of the piperidine ring.



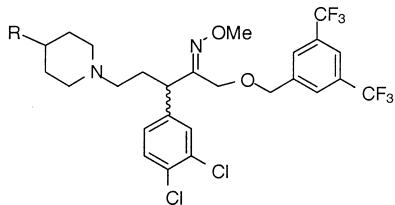
Results and Discussion

The substituted 4(*Z*)-(methoxyimino)pentyl-1-piperidines were prepared as the racemates by the synthetic route outlined in Scheme 1. Reductive amination⁶ of *N*-benzyl-4-piperidone (**2**) and subsequent hydrogenation⁷ provided amine **3**. Coupling of amine **3** with 1-[(3,5-bis(trifluoromethyl)phenyl)methoxy]-3-(3,4-dichlorophenyl)-5-formyl-2(*Z*)-pentanone *O*-methyl-oxime (**4**)⁸ produced the analogues **5**.⁹ The synthesis of compounds **9** began with the protection of 4-hydroxymethylpiperidine (**6**) as its *t*-BOC derivative. Swern oxidation¹⁰ of **6** followed by reductive amination and acid catalyzed removal of the *t*-BOC group afforded amine **8**. Addition of amine **8** to aldehyde **4** yielded the targets **9**.

The NK₁ and NK₂ biological activity of piperidines **10a–f** are reported in Table 1.¹¹ Replacement of the 4-hydroxy-4-phenylpiperidine subunit of **1** with the 4-(carbon linked amide)piperidines **10a** or **10b** increases NK₁ potency, but appears to decrease NK₂ potency.

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Table 1. NK₁ and NK₂ antagonistic activity of the piperidines **10a–f**



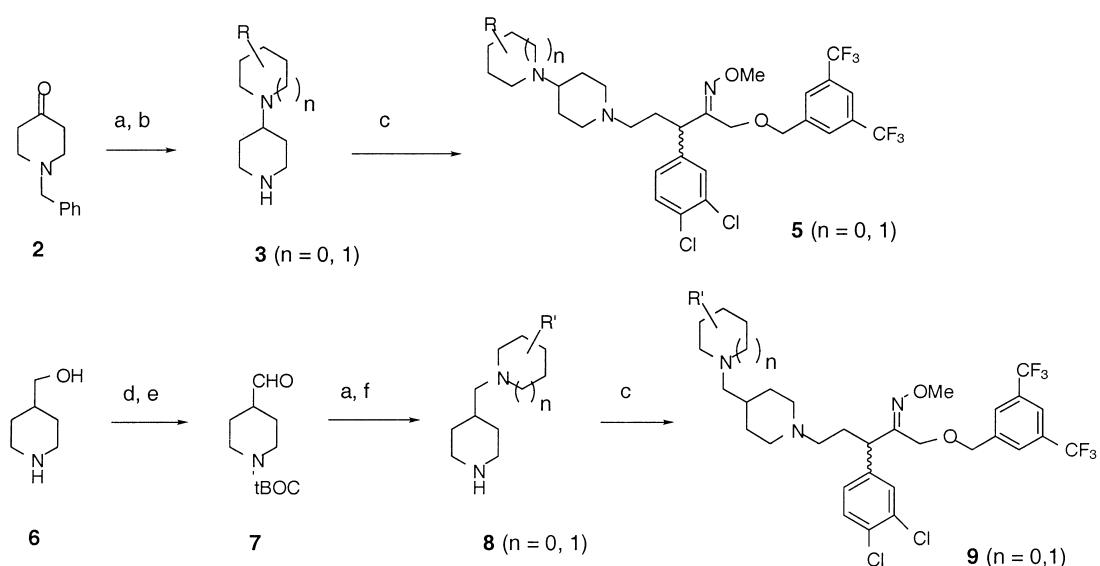
Compd	R	NK ₁ K _i (nM) ^a	NK ₂ K _i (nM) ^a
10a		8 (\pm 3)	67
10b		6 (\pm 2)	59
10c		18 (\pm 4)	144
10d		112	137
10e		44	146
10f		24 (\pm 7)	16 (\pm 8)

^aValues are means of two experiments. If values are means of three experiments, standard deviation is given in parentheses.

The 4-(nitrogen linked amide)piperidines **10c–e** show greatly reduced NK₂ activity. In this set of racemic analogues, only the hydroxyamino substituted piperidine **10f** is equipotent to our lead structure **1**.

Additional analogues that have a second piperidine or pyrrolidine ring attached to the 4-position of the piperidine are summarized in Table 2. The piperidone **11a**¹² shows increased potency for both the NK₁ and NK₂ receptors relative to the 4-hydroxy-4-phenylpiperidine **1**. Similarly, the corresponding pyrrolidone analogue **11f** and the thiopyrrolidine analogue **11g** both retain this desired dual NK₁/NK₂ profile. In contrast, the pyrrolidine analogue **11h** reduces NK₂ potency to a much greater extent than NK₁ potency. Introduction of a hydroxy moiety on the piperidine ring as in **11b** or the pyrrolidine ring as in **11i** retains NK₁ and NK₂ activity. A carboxyamide group can also be added to the piperidine ring as in **11d** or the pyrrolidine ring as in **11e** and produces potent NK₁ and NK₂ receptor affinity. However, a less polar carboxyester substituent as in **11c** or a carbamate substituent as in **11j** shows decreased biological activity.

In Table 3 our analogues that have a methylene linker between the two piperidine rings are tabulated. The parent piperidone **12a** and the pyrrolidone analogue **12d** possess increased NK₁ and NK₂ potency relative to our lead structure **1**. The thiopiperidone **12c** and the reduced piperidine analogue **12b** are slightly less active. Alkylation at the α -position of the piperidone **12a** gives analogues **12e–h**. Steric size appears to be detrimental to biological activity as evident by the benzyl analogue **12e**. Hydroxy groups are well-tolerated as in analogue **12h**. When the second piperidine ring is a morpholine ring as in **12i**, NK₁ activity is retained and NK₂ activity



Scheme 1. (a) Substituted amine, NaCNBH₃, CF₃CH₂OH, 3 Å sieves, 42–83%; (b) H₂, Pd/C, MeOH, 83–100%; (c) 1-[[3,5-bis(trifluoromethyl)phenyl]methoxy]-3-(3,4-dichlorophenyl)-5-formyl-2(Z)-pentanone *O*-methyloxime, 3 Å sieves, NaCNBH₃, CF₃CH₂OH, 32–63%; (d) (*t*-BOC)₂O, CH₂Cl₂, 100%; (e) ClCOCOCl, DMSO, CH₂Cl₂, Et₃N, 100%; (f) CF₃COOH, CH₂Cl₂, 100%.

decreases slightly. Substitution at the 2-position of the piperidine ring indicates that the hydroxyethyl side chain of **12k** is more active than the hydroxymethyl side chain of **12j**. Compounds **12l–p** vary the size and polarity of the moiety at the 3-position of the piperidine ring, and none of these analogues are as potent as the piperidone **12a**.

Table 2. NK₁ and NK₂ antagonistic activity of the piperidines **11a–j**

Compd	R	NK ₁ <i>K_i</i> (nM) ^a	NK ₂ <i>K_i</i> (nM) ^a
11a		18 (±4)	4 (±1)
11b		14 (±3)	15 (±6)
11c		117	20
11d		13 (±4)	10 (±3)
11e		13 (±3)	11 (±3)
11f		9 (±2)	7 (±2)
11g		16 (±3)	23 (±8)
11h		36	195
11i		4	16
11j		19 (±7)	67 (±19)

^aValues are means of two experiments. If values are means of three experiments, standard deviation is given in parentheses.

Table 3. NK₁ and NK₂ antagonistic activity of the piperidines **12a–p**

Compd	R	NK ₁ <i>K_i</i> (nM) ^a	NK ₂ <i>K_i</i> (nM) ^a
12a		10 (±3)	12 (±2)
12b		36 (±23)	26 (±4)
12c		20 (±7)	23 (±6)
12d		8 (±3)	21 (±6)
12e		69	91
12f		14 (±5)	26 (±7)
12g		13	58
12h		7 (±4)	16 (±8)
12i		14 (±10)	39 (±11)
12j		20	53
12k		13 (±3)	14 (±3)
12l		38	47
12m		50 (±30)	39 (±9)

(continued on next page)

Table 3 (continued)

Compd	R	NK ₁ K _i (nM) ^a	NK ₂ K _i (nM) ^a
12n		17 (±3)	33 (±5)
12o		25 (±10)	27 (±18)
12p		16 (±1)	28 (±7)

^aValues are means of two experiments. If values are means of three experiments, standard deviation is given in parentheses.

Conclusion

In conclusion, the left hand region of our lead NK₁/NK₂ inhibitor structure **1** appears to be quite tolerant to structural modifications. We have found that the 4-hydroxy-4-phenyl substituent of piperidine **1** can be replaced by a piperidone ring as in compounds **11a** and **12a**. From this structure–activity relationship study, several diverse analogues including compounds **11d**, **11e**, **11f**, and **12k** were found to be our most potent dual inhibitors. The results of our efforts to further optimize the biological profile of this series of dual NK₁/NK₂ antagonists will be forthcoming.

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